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MAY 22 2003  
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## STABLE AQUEOUS ANTIMICROBIAL SUSPENSION

### 5    **Related Application**

[0001] This application claims benefit of Provisional U.S. Application No. 60/264,611 filed January 27, 2001.

### 10    **Technical Field**

[0002] This invention relates to antimicrobial compositions, and particularly to aqueous-based compositions which are stable and effective over long periods.

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### **Background of the Invention**

[0003] Haloacetamides are used extensively as antimicrobial agents in various industrial applications, such as water treatment and  
20    preservation. The active ingredient (the haloacetamide) is a solid, which is difficult to feed in industrial applications and poses problems in material handling. Because of the problems in handling solids, liquid concentrates have been developed. Such liquid concentrates are convenient for their ability to be diluted, and their relative ease of  
25    application.

[0004] While it is desirable to make and use haloacetamides in liquid form, it has been difficult to formulate a stable aqueous formulation. Haloacetamides decompose rapidly by hydrolysis or photolysis. Moreover, most suspending agents tend to break down under acidic conditions. Currently used commercial formulations utilize a mixture of organic solvents and water, or, because of the proclivity of the haloacetamide to hydrolyze, sometimes the solvent without water, to carry the haloacetamides. Users have raised concerns about the organic solvents because of their toxicity to man by occupational exposure and to the environment.

[0005] Xanthate gum has been proposed for use as a thixotropic suspending agent for suspensions of 2,2-dibromo-3-nitrilopropionamide (DBNPA) by Gartner in US Patent 5,627,135. However, Miskiel and Solanki, in US Patent 6,083,890, have shown that acidic cleaning compositions containing xanthan gum and a preservative (5-bromo-5-nitro-1,3-dioxane) rapidly lose viscosity, while a low-acetate xanthan gum maintained the viscosity stability or even increased it. See Table 1 of US Patent 6,083,890. The natural xanthan gum, containing at least 5% acetic acid groups, typically 5.6% by weight, itself degrades in an acidic environment. As reviewed by Miskiel and Solanki column 3, lines 33-47, "Although xanthan gum is well known as a rheology modifier in cleansers, characteristically the viscosity decreases undesirably over time at low pH, within about seven days after making the compositions. The

extent to which the viscosity decreases is dependent on a number of factors, such as the pH and ionic strength of the cleaner and the pH levels, and the temperature of the acidic cleaner composition at which  
5 it is stored. In compositions stored at ambient temperature, xanthan gum loses a significant proportion, perhaps greater than about 20% or more, of its viscosifying functionality within an acidic composition in about seven days at a pH of about 2.2 or less. This may eventually lead to product performance disappointment and failure unless an  
10 increased concentration of xanthan gum is initially used to compensate for the decrease in viscosity.”

[0006] The difficulty of creating a stable suspension of a haloacetamide with xanthan gum is compounded by the fact, as  
15 mentioned above, that the haloacetamides tend to hydrolyze in water and especially so at higher pH's. Thus the desirability of a low pH to preserve the haloacetamide conflicts with the adverse effects of a low pH on a suspending agent such as natural xanthan gum. Nevertheless, Gartner, in US Patent 5,627,135, recommends reducing  
20 the pH of the water to below 7 before adding the natural xanthan and says that “the pH of the formulation will usually equilibrate to about 1 to about 4 and no further acidification is needed.” Col 5 lines 34-51. His Table 1, however, contains no examples using xanthan gum alone as the suspending agent.

[0007] An acid stable liquid formulation of a haloacetamide is needed in the industry. The need is especially critical for a stable formulation of 2,2 dibromo 3-nitrilopropionamide ("DBNPA").

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## SUMMARY OF THE INVENTION

[0008] This invention includes a formulation of an aqueous suspension or dispersion of haloacetamide that only uses water as the solvent and is stable when stored. The invention uses a unique agent capable of suspending haloacetamides over a broad range of concentration, inhibiting hydrolysis. The haloacetamides are preferably suspended in concentrations from 5% to 60% by weight, although higher concentrations can be used where high viscosities can be tolerated.

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[0009] To suspend the formulations, an acetate-free xanthan gum is used in a concentration ranging from 0.1% to 5%, anchoring the pH between 1 and 5 with a buffer comprising sodium acetate and acetic acid in a weight ratio of 1.5:1 to 2.5:1, in an amount effective to maintain the pH between 1 and 5 for a desired period of stability. The invention provides:

- a. Storage Stability equivalent to other commercially available solutions.
- b. Equivalent microbiological efficacy to other commercially available formulations over the use of the formulation.

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haloacetamide (preferably 5% to 60%, more preferably 10% to 45% and most preferably 15% to 25% by weight), 0.1% to 5% by weight (preferably 0.5% to 4%) of an acetate-free xanthan gum suspending agent, and acetic acid, sodium acetate or a mixture thereof as a buffering agent effective to maintain the suspension at a pH between 1 and 5, preferably between 3.8 and 4.2. Typically, an effective amount of buffering agent will comprise 1-2% sodium acetate and 0.5-1% acetic acid, preferably in a weight ratio of 1.5:1 to 2.5:1. Our invention includes a method of making the suspension, comprising forming an aqueous solution of 0.1% to 5% by weight of an acetate-free xanthan gum, adding the buffer, and then adding the haloacetamide in the proportions desired to make a composition as described above. The buffer is added not merely to reduce the initial pH (cf Gartner US Patent 5,627,135 col 5 lines 34-50) but to maintain it over a period of time to inhibit hydrolysis of the DBNPA.

Our invention is applicable to any of the halogenated amides recited in Burk et al US Patent 4,163,798, which is incorporated herein by reference in its entirety. In particular, the halogenated amides useful in our invention are alpha-haloamides; that is, compounds which contain an amide functionality [ie a moiety of the formula  $-C(O)-N<$ ] and which have at least one halogen atom on a carbon atom located adjacent to (that is, in the alpha position relative to) the carbonyl group  $[-C(O)-]$  of such amide functionality. Preferably, they are halogenated nitrilopropionamides. Examples of the preferred group

are 2,2 dibromo 3-nitrilopropionamide ("DBNPA"), 2-bromo-2-cyano-N,N-dimethylacetamide, 2-bromo 3-nitrilopropionamide, 2-bromo 2,3-dinitrilopropionamide, N,N-dimethyl-2,2-dibromo-3-nitrilopropionamide, and N-(n-propyl)-2-iodo-2bromo-3-nitrilopropionamide. A most preferred haloacetamide is 2,2 dibromo 3-nitrilopropionamide ("DBNPA"). A preferred buffering agent comprises sodium acetate and acetic acid, preferably in a molar ratio of 1.5:1 to 2.5: 1, and more preferably about 2:1.

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Suspensions and/or dispersions of the above described formulations are stable and effective over long periods of time, are conveniently prepared and dispensed for use, and are more acceptable environmentally and with respect to toxicity than comparable conventional antimicrobial compositions.

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### **Brief Description of the Drawings**

**Figure 1** shows graphically the known rate of hydrolysis of DBNPA at 25°C.

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### **Detailed Description of the Invention**

Figure 1 is a plot of the known hydrolysis in water of DBNPA. It will be seen that the lowest rate of hydrolysis is at slightly less than pH 4.

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